

# Chapter 19 Acids Bases Salts Work Answers

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Chapter 19 Acids Bases Salts

Chapter 19 Acids, Bases, and Salts. Test on Thursday May 10, 2012.

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Acids taste sour, will change the color of an acid-base indicator, and can be strong or weak electrolytes in aqueous solution. Bases taste bitter, feel slippery, will change the color of an acid-base indicator, and can be strong or weak electrolytes in aqueous solution.

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Strong Acid + Strong Base = Neutral Salt  
Strong Acid + Weak Base = Acidic Salt  
Weak Acid + Strong Base = Basic Salt  
Weak Acid + Weak Base = any type depending on the relative strengths of the acid and base involved.

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Chapter 19...Acids, Bases, and Salts ACID-BASE THEORIES Acids and bases

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are all around us and part of our everyday life (ex. bodily functions, vinegar, carbonated drinks, citrus fruits, car batteries, antacids, cleaning products, etc.). Acids and bases have the following properties: Acids Bases Tart or sour taste Bitter taste

Chapter 19 Acids, Bases, and Salts - Science with Mrs ...

Arrhenius said that acids are hydrogen-containing compounds that ionize to yield hydrogen ions ( $H^+$ ) in aqueous solution. He also said that bases are compounds that ionize to yield hydroxide ions ( $OH^-$ ) in aqueous solution.

Chapter 19 Acids, Bases, and Salts Flashcards

The Acids, Bases and Salts chapter of this Prentice Hall Chemistry Companion Course helps students learn the essential lessons associated with acids, bases and salts.

Prentice Hall Chemistry Chapter 19: Acids, Bases and Salts ...

Acids and bases come in pairs. General equation is:  $HA(aq) + H_2O(l) \rightleftharpoons H_3O^+(aq) + A^-(aq)$  Acid + Base  $\rightleftharpoons$  Conjugate acid + Conjugate base.  $NH_3 + H_2O \rightleftharpoons NH_4^+ + OH^-$  1-base acid c.a. c.b.  $HCl + H_2O \rightleftharpoons H_3O^+ + Cl^-$  1-acid base c.a. c.b. Amphoteric – a substance that can act as both an acid and base- as water shows

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Prentice Hall Chemistry Chapter 19: Acids, Bases and Salts ...

Chapter 19 – Acids, Bases, and Salts. Jennie L. Borders. Section 19.1 – Acid-Base Theories. Acids have a sour taste, change the color of an indicator, can be strong or weak electrolytes in aqueous solution, and react with metals. Bases.

Chapter 19 – Acids, Bases, and Salts

596 Chapter 19 Acids and Bases PRACTICE PROBLEMS 1. Write a balanced formula equation for the reaction that occurs between each of the

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following pairs of reactants. a. magnesium and nitric acid b. aluminum and sulfuric acid c. calcium carbonate and hydrobromic acid d. potassium hydrogen carbonate and hydrochloric acid

Chapter 19: Acids and Bases

Chapter 10 - moles (handouts) Chapter 11 - reactions (handouts)

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mail to a friend; Find other activities ... compounds derived from the reaction of a strong base with a weak acid or a strong acid with a weak base or a solution of a weak base and one of its salts ...

solution that consists of a weak acid and one of its salts: acid ...

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Chapter 19 - Acids, Bases, and Salts - 19 Assessment ...

Acids, Bases and Salts Notes of CBSE Class 10 Science Chapter with detailed explanation of the chapter 'Acids, bases and salts' along with meanings of difficult words. Given here is the complete explanation of the chapter, along with examples and all the exercises, Question and Answers given at the back of the chapter.

Acids, Bases and Salts Class 10 Science Chapter ...

Chapter 19 Acids, Bases, and Salts - Chapter 19 Acids, Bases, and Salts Buffers & Your Blood Your body function properly only when the pH of your blood lies between 7.35 and 7.45.

PPT – Chapter 19 Acids, Bases, and Salts PowerPoint ...

Unformatted text preview: Chapter 19 "Acids, Bases, and Salts"  
Chemistry Section 19.1 Acid-Base Theories OBJECTIVES: –Define the properties of acids and bases. Section 19.1 Acid-Base Theories OBJECTIVES: –Compare and contrast acids and bases as defined by the theories of: a) Arrhenius, b) Brønsted-Lowry, and c) Lewis.

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Chemistry - Chapter 19 Acids, Bases, and Salts - Chapter ...

Acids: taste sour, react with certain metals to produce hydrogen gas, react with carbonates/bicarbonates to produce carbon dioxide gas. Corrode metals, pH is less than 7, turns blue litmus paper red.

Bases: taste bitter, have a slippery feeling. React with acids to form salts and water, pH is greater than 7.

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